Assignment for SSC Examinees, 2021

Subject: Chemistry

Subject Code: 137

Level: SSC

Assignment	Assignment	Learning	Guidelines		Ass	sessment Criteri	ion /Rubric			Com't
Number,		Outcomes	(cues/steps or							
Chapter		1	stages)							
Number,										
Chapter Title	>	Ct 1 t-2 i11	TT 4 . C 1			D. (:				
01	Number of neutrons in	Students' will	•Has to find	Indicaton		Kating	g Scale		Score	
01	four different elements, Figure of structure of	be ● Able to	out the number of neutrons in	Indicator	1	3	2	1		
Chapter	atom according to the	determine the	four elements		Has found out	Has found out	Has found out	Has found out		
Three:	Bhor's model,	number of	Total Cicincias	a) Calculation	the correct	the correct	the correct	the correct		
	Electronic	electrons,	Has to draw	of neutron	number of	number of	number of	number of		
Structure of	configuration of energy	protons and	the figure of	number	neutrons of	neutrons of 3	neutrons of 2	neutrons of 1		
Matter	level and sub-energy	neutrons in an	structure of		four isotopes	isotopes	isotopes	isotope		
	level(orbitals) Prepare a report on the number of neutrons in the mass numbers of the elements mentioned next	Able to describe the structure of atom in relation to the theories of Rutherford and Bohr atomic model Able to write the electronic configurations of different orbits and different sub levels of orbits of an	atom according to the Bhor's model • Has to write the electronic	b)Drawing of structure of atom according to Bhor's model	Has drawn the exact figure of four structure of atoms according to Bhor's model	Has drawn the exact figure of three structure of atoms according to Bhor's model	Has drawn the exact figure of two structure of atoms according to Bhor's model	Has drawn the exact figure of one structure of atom according to Bhor's model		
	diagram of the structure of atoms according to Bohr model, their electronic configuration at energy levels and sub- energy (orbitals) levels.		of energy level of four elements • Has to write the electronic	c) Electronic configuration of an energy level	Has written the electronic configurations of energy level of four elements correctly	Has written the electronic configurations of energy level of three elements correctly	Has written the electronic configurations of energy level of two elements correctly	Has written the electronic configuration of energy level of one element correctly		
	Na(11), Mass number-23 P(15), Mass number-31 K(19), Mass number-40 Cu(29), Mass number-63		configurations of sub-energy level (orbitals) of four elements	d)Sub- energy level(orbitals) electronic configuration	Has written the electronic configurations of sub-energy level of four elements correctly	Has written the electronic configurations of sub- energy level of three elements correctly	Has written the electronic configurations of sub-energy level of two elements correctly	Has written the electronic configuration of sub- energy level of one element correctly		
		atom		T-t-1 moules for	41.i. a saignmand	1.10		Total		
				Total marks for	r this assignment:	: 16				

Marks Obtained	Comments			
13-16	Excellent			
11-12	Very good			
08-10	Good			
0-07	Needs improvement			